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**IMPROVED INK JET RECORDING MATERIAL****CROSS-REFERENCE TO RELATED PATENT APPLICATIONS**

5 This application claims the benefit of U.S. Provisional  
Application No. 60/421,656 filed October 28, 2002, which is  
incorporated by reference. In addition, this application claims  
the benefit of European Application No. 02102448.4 filed October  
15, 2002, which is also incorporated by reference.

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**FIELD OF THE INVENTION**

The present invention relates to an ink jet recording  
material having an improved stabilization of the finished image  
15 against color fading due to light.

**BACKGROUND OF THE INVENTION**

In the majority of applications printing proceeds by pressure  
20 contact of an ink-laden printing form with an ink-receiving  
material which is usually plain paper. The most frequently used  
impact printing technique is known as lithographic printing based  
on the selective acceptance of oleophilic ink on a suitable  
receptor.

25 In recent times however so-called non-impact printing systems  
have replaced classical pressure-contact printing to some extent  
for specific applications. A survey is given e.g. in the book  
"Principles of Non Impact Printing" by Jerome L. Johnson (1986),  
Palatino Press, Irvine, CA 92715, USA.

30 Among non-impact printing techniques ink jet printing has  
become a popular technique because of its simplicity, convenience  
and low cost. Especially in those instances where a limited  
edition of the printed matter is needed ink jet printing has  
become a technology of choice. A recent survey on progress and  
35 trends in ink jet printing technology is given by Hue P. Le in  
Journal of Imaging Science and Technology Vol. 42 (1), Jan/Febr  
1998.

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In ink jet printing tiny drops of ink fluid are projected directly onto an ink receptor surface without physical contact between the printing device and the receptor. The printing device stores the printing data electronically and controls a mechanism for ejecting the drops image-wise. Printing is accomplished by moving the print head across the paper or vice versa. Early patents on ink jet printers include US 3,739,393, US 3,805,273 and US 3,891,121.

The jetting of the ink droplets can be performed in several different ways. In a first type of process, a continuous droplet stream is created by applying a pressure wave pattern. This process is known as continuous ink jet printing. In a first embodiment the droplet stream is divided into droplets that are electrostatically charged, deflected and recollected, and into droplets that remain uncharged, continue their way undeflected, and form the image. Alternatively, the charged deflected stream forms the image and the uncharged undeflected jet is recollected. In this variant of continuous ink jet printing, several jets are deflected to a different degree and thus record the image (multideflection system).

According to a second process the ink droplets can be created "on demand" ("DOD" or "drop on demand" method) whereby the printing device ejects the droplets only when they are used in imaging on a receiver thereby avoiding the complexity of drop charging, deflection hardware, and ink recollection. In drop-on-demand the ink droplet can be formed by means of a pressure wave created by a mechanical motion of a piezoelectric transducer (so-called "piezo method"), or by means of discrete thermal pushes (so-called "bubble jet" method, or "thermal jet" method).

Ink compositions for ink jet typically include following ingredients : dyes or pigments, water and/or organic solvents, humectants such as glycols, detergents, thickeners, polymeric binders, preservatives, etc. It will be readily understood that the optimal composition of such an ink is dependent on the ink jetting method used and on the nature of the substrate to be printed. The ink compositions can be roughly divided in :

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- water based ; the drying mechanism involves absorption, penetration and evaporation;

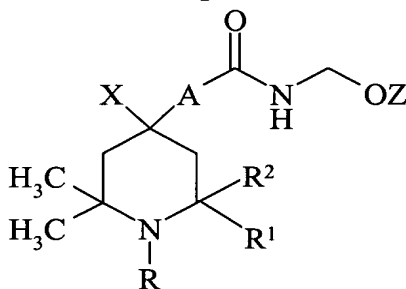
- oil based ; the drying involves absorption and penetration;

- solvent based ; the drying mechanism involves primarily evaporation;

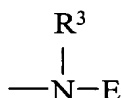
- hot melt or phase change : the ink vehicle is liquid at the ejection temperature but solid at room temperature ; drying is replaced by solidification;

- UV-curable ; drying is replaced by polymerization.

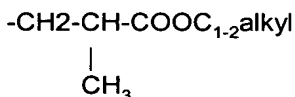
US 4,223,147 discloses a compound of the formula



wherein A is



wherein  $R^3$  is  $C_1$ - $C_{20}$ -alkyl, hydrogen,  $C_5$ - $C_{12}$ -cycloalkyl or  $C_7$ - $C_{12}$ -aralkyl and E is a direct bond; Z is hydrogen or  $C_1$ - $C_{14}$ -alkyl; R is hydrogen,  $C_1$ - $C_{20}$ -alkyl,  $C_3$ - $C_5$ -alkenyl;  $C_7$ - $C_{12}$ -aralkyl;  $CH-CHR^6-OH$ , wherein  $R^6$  is hydrogen, methyl, or phenyl;  $-CH_2CH_2CN$ ;  $-CH_2-CH_2-COOC_{1-2}alkyl$ ; or



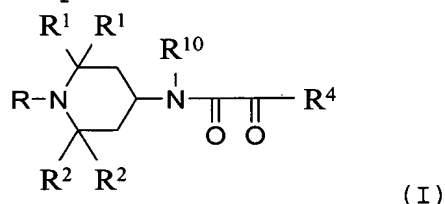
$R^1$  and  $R^2$  independently of one another are  $C_1$ - $C_6$ -alkyl, or  $R^1$  and  $R^2$  together with the ring carbon atom to which they are attached form  $C_1$ - $C_7$ -cycloalkyl; and X is hydrogen or methyl. US 4,223,147 further discloses a process for the production of this compound.

US 4,730,017 discloses a method of stabilizing a lacquer composition based on a member selected from the group consisting of acrylic resins, alkyl resins, polyester resins and acrylic

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resins, alkyd resins and polyester resins crosslinked with melamine/formaldehyde resins, epoxide resins or polyisocyanates, which comprises incorporating into the resin a light-stabilizing-effective amount of at least one 4-oxalamido-2,2,6,6-

5 tetraalkylpiperidine compound of formula I



in which R is hydrogen, oxygen, C<sub>1-8</sub> alkyl or -CO-R<sup>5</sup> ; each R<sup>1</sup>, independently, is -CH<sub>3</sub> or -CH<sub>2</sub> (C<sub>1-4</sub>alkyl) or both groups R<sup>1</sup> form a  
10 group -CH<sub>2</sub>)<sub>5</sub> ; each R<sup>2</sup>, independently, is -CH<sub>3</sub> or -CH<sub>2</sub> (C<sub>1-4</sub>alkyl) or both groups R<sup>2</sup> form a group -CH<sub>2</sub>)<sub>5</sub> ; R<sup>4</sup> is an amide-forming group;

R<sup>5</sup> is -C(R<sup>10</sup>)=CH<sub>2</sub>, C<sub>1-6</sub>alkyl, phenyl, -CO-O-C<sub>1-4</sub>alkyl or -NR<sup>7</sup>R<sup>8</sup> ;  
R<sup>7</sup> is hydrogen, C<sub>1-2</sub>alkyl, C<sub>5-6</sub>cycloalkyl, phenyl, phenyl C<sub>1-4</sub>alkyl  
15 or C<sub>1-12</sub>alkylphenyl; R<sup>8</sup> is C<sub>1-12</sub>alkyl or hydrogen; and R<sup>10</sup> is hydrogen or C<sub>1-4</sub>alkyl. US 4,730,017 further discloses a lacquer composition based on a member selected from the group consistency of acrylic resins, alkyl resins, polyester resins and acrylic resins, alkyd resins and polyester resins crosslinked with  
20 melamine/formaldehyde resins, epoxide resins or polyisocyanates, which contains at least one 4-oxalamido-2,2,6,6-tetraalkylpiperidine compound of formula I.

JP 2000-211237A discloses an ink-jet image recording medium characterized by preparing one or more layers containing polyvinyl  
25 alcohol on a base material apart from one or more layers which contain at least one sort of polymer which has (a) alkylene oxide chain as the ink fixed bed, and the (b) above-mentioned (a) layer. JP 2000-211237A further discloses an ink jet recording method using the claimed image recording medium.

30 In the unpublished patent application EP 03102692.5, with an earlier priority date than the present application, an ink jet recording material is disclosed comprising a support and at least one binder containing ink-receiving layer, characterized in that

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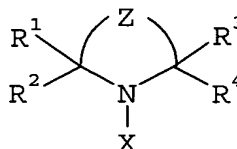
said at least one ink-receiving layer further contains a compound according to following general formula (I):



5

wherein,

- A is represented by following formula:



10

wherein :

Z represents the necessary atoms to complete a five- or six-membered ring,

R<sup>1</sup> to R<sup>4</sup> independently represent a substituted or unsubstituted C1 to C6 aliphatic group,

15

X is selected from the group consisting of a hydrogen, a substituted or unsubstituted aliphatic group, an acyl group, an oxy radical, a hydroxyl group, an alkoxy group, an -OSO<sub>2</sub>-alkyl group, and an acyloxy group;

20

- L is a divalent linking group, linked to the five- or six-membered ring by one of the atoms of Z, optionally by a double bond, characterized in that said divalent linking group comprises a nitrogen-nitrogen or nitrogen-oxygen bond,

- R represents a non aromatic moiety comprising at least two hydroxyl groups.

25

It is known that the ink-receiving layers in ink-jet recording elements must meet different stringent requirements:

30

- the ink-receiving layer should have a high ink absorbing capacity, so that the dots will not flow out and will not be expanded more than is necessary to obtain a high optical density;

- the ink-receiving layer should have a high ink absorbing speed (short ink drying time) so that the ink droplets will not feather if smeared immediately after applying;

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- the ink dots that are applied to the ink-receiving layer should be substantially round in shape and smooth at their peripheries. The dot diameter must be constant and accurately controlled;
- the receiving layer must be readily wetted so that there is no "puddling", i.e. coalescence of adjacent ink dots, and an earlier absorbed ink drop should not show any "bleeding", i.e. overlap with neighbouring or later placed dots;
- transparent ink-jet recording elements must have a low haze-value and be excellent in transmittance properties;
- after being printed the image must have a good resistance regarding water-fastness, light-fastness, and good endurance under severe conditions of temperature and humidity;
- the ink jet recording element may not show any curl or sticky behaviour if stacked before or after being printed;
- the ink jet recording element must be able to move smoothly through different types of printers;

All these properties are often in a relation of trade-off. It is difficult to satisfy them all at the same time. A particular problem is the stability of the color densities of the finished color ink jet image when exposed to light for a longer period ("light-fastness"). As well-known by those skilled in the art, the light fading of colorants is mainly due to an oxidative decomposition of the colorant catalyzed by light, in particular by the UV spectral part. Therefore, there is a permanent need of more effective compounds, which stabilize the colorants of the ink jet image against fading by light.

#### OBJECTS OF THE INVENTION

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It is therefore an object of the present invention to provide an ink jet recording material with an improvement in light-fastness.

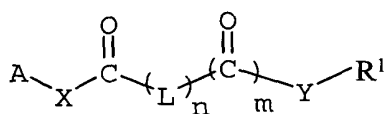
It is therefore a further object of the present invention to provide a compound having light-stabilizing properties.

Further objects and advantages of the present invention will become apparent from the description hereinafter.

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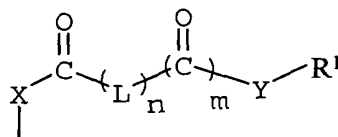
## SUMMARY OF THE INVENTION

It has been surprisingly found that when an ink jet recording material contains a compound represented by formula (I):



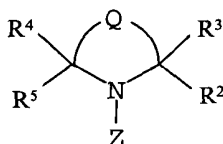
formula (I)

wherein



is exclusive of a nitrogen-nitrogen or nitrogen-oxygen bond, and

wherein A is represented by following formula :

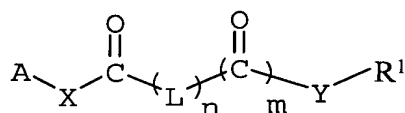


wherein,

Q represents the necessary atoms to complete a five- or six-membered ring; R<sup>2</sup> to R<sup>5</sup> independently represent a substituted or unsubstituted C1 to C6 aliphatic group; Z is selected from the group consisting of hydrogen, a substituted or unsubstituted aliphatic group, an acyl group, an oxy radical, a hydroxyl group, an alkoxy group and an acyloxy group; L is a divalent linking group linked to a carbonyl group; X and Y are independently selected from an oxygen and NR<sup>6</sup>, wherein R<sup>6</sup> is selected from the group consisting of hydrogen, a substituted or unsubstituted, saturated or unsaturated aliphatic group, a substituted or unsubstituted aromatic group, and a substituted or unsubstituted

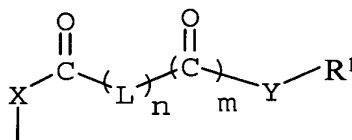
heteroaromatic group; X is linked to A via one of the atoms of Q; R<sup>1</sup> represents a non-aromatic moiety comprising at least two hydroxyl groups; n and m independently represent 1 or 0, exhibit a strong improvement in light-fastness.

Objects of the present invention are realized by providing an ink jet recording material comprising a support and at least one binder-containing ink-receiving layer, characterized in that said ink-receiving layer further contains a light-stabilizing compound represented by formula (I):

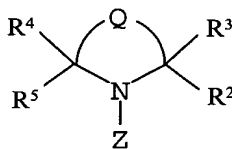


formula (I)

wherein



is exclusive of a nitrogen-nitrogen or nitrogen-oxygen bond, and wherein A is represented by following formula :



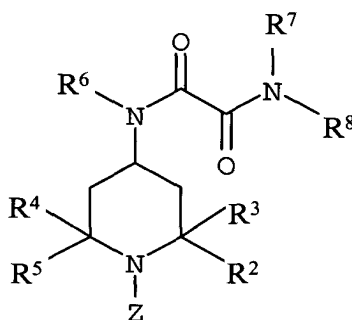
wherein,

Q represents the necessary atoms to complete a five- or six-membered ring; R<sup>2</sup> to R<sup>5</sup> independently represent a substituted or unsubstituted C1 to C6 aliphatic group; Z is selected from the group consisting of hydrogen, a substituted or unsubstituted aliphatic group, an acyl group, an oxy radical, a hydroxyl group, an alkoxy group and an acyloxy group; L is a divalent linking group linked to a carbonyl group; X and Y are independently selected from an oxygen and NR<sup>6</sup>, wherein R<sup>6</sup> is selected from the

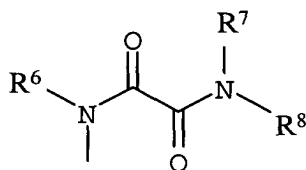


group consisting of hydrogen, a substituted or unsubstituted, saturated or unsaturated aliphatic group, a substituted or unsubstituted aromatic group, and a substituted or unsubstituted heteroaromatic group; X is linked to A via one of the atoms of Q;  
 5 R<sup>1</sup> represents a non-aromatic moiety comprising at least two hydroxyl groups; and n and m independently represent 1 or 0.

Objects of the present invention are preferably realized by providing a compound according to following formula:



10 wherein,



is exclusive of a nitrogen-nitrogen or nitrogen-oxygen bond, and  
 15 wherein

R<sup>2</sup> to R<sup>5</sup> independently represent a substituted or unsubstituted C1 to C6 aliphatic group;

Z is selected from the group consisting of hydrogen, a substituted or unsubstituted aliphatic group, an acyl group, an oxy radical, a  
 20 hydroxyl group, an alkoxy group and an acyloxy group; R<sup>6</sup> is

selected from the group consisting of hydrogen, a substituted or unsubstituted, saturated or unsaturated aliphatic group, a substituted or unsubstituted aromatic group, and a substituted or unsubstituted heteroaromatic group; R<sup>7</sup> represents a non-aromatic

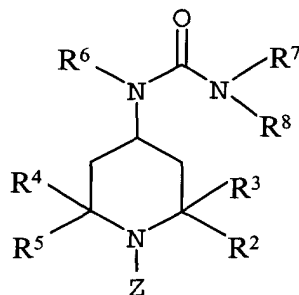
25 moiety comprising at least two hydroxyl groups; R<sup>8</sup> is selected from the group consisting of hydrogen, substituted or

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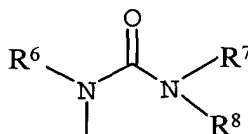
unsubstituted, saturated or unsaturated aliphatic group, a substituted or unsubstituted aromatic group.

Objects of the present invention are preferably realized by providing a compound according to following formula:

5



wherein,



is exclusive of a nitrogen-nitrogen or nitrogen-oxygen bond, and  
 10 wherein  
 $R^2$  to  $R^5$  independently represent a substituted or unsubstituted C1 to C6 aliphatic group;  
 Z is selected from the group consisting of hydrogen, a substituted or unsubstituted aliphatic group, an acyl group, an oxy radical, a  
 15 hydroxyl group, an alkoxy group and an acyloxy group;  $R^6$  is selected from the group consisting of hydrogen, a substituted or unsubstituted, saturated or unsaturated aliphatic group, a substituted or unsubstituted aromatic group, and a substituted or unsubstituted heteroaromatic group;  $R^7$  represents a non-aromatic  
 20 moiety comprising at least two hydroxyl groups;  $R^8$  is selected from the group consisting of hydrogen, substituted or unsubstituted, saturated or unsaturated aliphatic group, a substituted or unsubstituted aromatic group.

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Further advantages and embodiments of the present invention will become apparent from the following description.

5 DETAILED DESCRIPTION OF THE INVENTION

Definitions

The term divalent linking group as used in disclosing the  
10 present invention refers to a linkage group linking two entities by covalent bonds, regardless of whether the linkage group is linked to each of these entities via a single, double or triple bond.

The term alkyl means all variants possible for each number of  
15 carbon atoms in the alkyl group i.e. for three carbon atoms: n-propyl and isopropyl; for four carbon atoms: n-butyl, isobutyl and tertiary-butyl; for five carbon atoms: n-pentyl, 1,1-dimethylpropyl, 2,2-dimethylpropyl and 2-methyl-butyl etc.

The term acyl group as used in disclosing the present  
20 invention means  $-(C=O)-$ aryl and  $-(C=O)-$ alkyl groups.

The term saturated aliphatic group as used in disclosing the present invention means saturated straight chain, branched chain and alicyclic hydrocarbon groups.

The term unsaturated aliphatic group as used in disclosing  
25 the present invention means straight chain, branched chain and alicyclic hydrocarbon groups which contain at least one double or triple bond.

The term aromatic group as used in disclosing the present invention means an assemblage of cyclic conjugated carbon atoms,  
30 which are characterized by large resonance energies, e.g. benzene, naphthalene and anthracene.

The term non-aromatic moiety as used in disclosing this invention means any group which is not an aromatic group as defined above but includes heteroaromatic groups. Examples of non-  
35 aromatic moieties are saturated aliphatic groups, unsaturated aliphatic groups and an assemblage of cyclic conjugated carbon atoms, which are not characterized by large resonance energies.

One or more carbon atoms in said assemblage of cycl c conjugated carbon atoms of said non-aromatic moiety, may be replaced by an atom selected from the group of oxygen, nitrogen, sulfur, selenium and tellurium.

15 Ink-recording material

20 Support

The support for use in the present invention can be chosen from paper type and polymeric type supports well-known from photographic technology. Paper types include plain paper, cast coated paper, polyethylene coated paper and polypropylene coated paper. Polymeric supports include cellulose acetate propionate or cellulose acetate butyrate, polyesters such as polyethylene terephthalate and polyethylene naphthalate, polyamides, polycarbonates, polyimides, polyolefins, poly(vinylacetals), polyethers and polysulfonamides. Other examples of useful high-quality polymeric supports for the present invention include opaque white polyesters and extrusion blends of polyethylene terephthalate and polypropylene. Polyester film supports and especially poly(ethylene terephthalate) are preferred because of their excellent properties of dimensional stability. When such a polyester is used as the support material, a subbing layer may be

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employed to improve the bonding of the ink-receiving layer to the support. Useful subbing layers for this purpose are well known in the photographic art and include, for example, polymers of vinylidene chloride such as vinylidene chloride/acrylonitrile/  
5 acrylic acid terpolymers or vinylidene chloride/methyl acrylate/itaconic acid terpolymers.

#### Ink-receiving layer and optional auxiliary layers

10 The ink-receiving layer contains, apart from a binder, a light-stabilizer according to formula (I) as disclosed above. The light-stabilizers are preferably added to the coating solution of the ink-receiving layer as aqueous solutions. The amount of the light-stabilizer in the receiving layer is preferably comprised  
15 between 0.5 and 3 g/m<sup>2</sup>. A mixture of two or more light-stabilizers may be used. Mixtures of two or more binders can also be used in the receiving layer.

The ink-receiving layer may consist of just one single layer, or alternatively it may be composed of two layers or even of  
20 multiple layers. A particular type of an extra top ink-receiving layer may be designated as a so-called "gloss improving layer", meaning a layer which achieves a gloss of more than 30 at a 60° angle. This gloss property can be achieved by the use of swellable polymers and/or (in)organic pigments with a particle size smaller  
25 than 500 nm.

In the case of double or multiple ink-receiving layers the light-stabilizer may be incorporated in just one layer, or in several layers or in all layers. It may also be present in additional auxiliary layers, if present, such as an anti-curl  
30 backing layer.

The ink-receiving layer or in at least one of the ink-receiving layers, in the case of multiple layers, according to the present invention, may further contain a pigment.

35 The ink-receiving layer or in at least one of the ink-receiving layers, in the case of multiple layers, according to the present invention, may also further contain a cationic substance acting as mordant.

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The ink-receiving layer, and an optional auxiliary layer, such as a backing layer for anti-curl purposes, may further contain well-known conventional ingredients, such as surfactants serving as coating aids, hardening agents, plasticizers, whitening agents and matting agents.

The ink-receiving layer and the optional auxiliary layer(s) may also be crosslinked to provide such desired features as waterfastness and non-blocking characteristics. The crosslinking is also useful in providing abrasion resistance and resistance to the formation of fingerprints on the element as a result of handling.

The different layers can be coated onto the support by any conventional coating technique, such as dip coating, knife coating, extrusion coating, spin coating, slide hopper coating and curtain coating.

#### Binder

The binder can be chosen from a list of compounds well-known in the art including hydroxyethyl cellulose; hydroxypropyl cellulose; hydroxyethylmethyl cellulose; hydroxypropyl methyl cellulose; hydroxybutylmethyl cellulose; methyl cellulose; sodium carboxymethyl cellulose; sodium carboxymethylhydroxyethyl cellulose; water soluble ethylhydroxyethyl cellulose; cellulose sulfate; polyvinyl alcohol; vinylalcohol copolymers; polyvinyl acetate; polyvinyl acetal; polyvinyl pyrrolidone; polyacrylamide; acrylamide/acrylic acid copolymer; polystyrene, styrene copolymers; acrylic or methacrylic polymers; styrene/acrylic copolymers; ethylene-vinylacetate copolymer; vinyl-methyl ether/maleic acid copolymer; poly(2-acrylamido-2-methyl propane sulfonic acid); poly(diethylene triamine-co-adipic acid); polyvinyl pyridine; polyvinyl imidazole; polyethylene imine epichlorohydrin modified; polyethylene imine ethoxylated; polyethylene oxide; polyurethane; melamine resins; gelatin; carrageenan; dextran; gum arabic; casein; pectin; albumin; starch; collagen derivatives; collodion and agar-agar.

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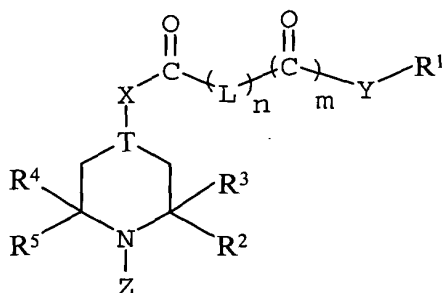
A preferred binder for the practice of the present invention is a polyvinylalcohol (PVA), a vinylalcohol copolymer or modified polyvinyl alcohol. Most preferably, the polyvinyl alcohol is a cationic type polyvinyl alcohol, such as the cationic polyvinyl alcohol grades from Kuraray, such as POVAL C506, POVAL C118, and from Nippon Goshei.

Compound represented by formula (I)

According to a preferred embodiment of the ink-recording material, according to the present invention, Z is selected from the group consisting of a hydrogen atom and  $-O^-$ ,  $-OH$  and methoxy groups.

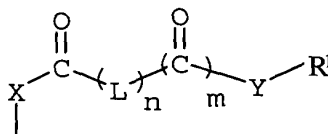
According to a preferred embodiment of the ink-recording material, according to the present invention, Q is selected from the group consisting of optionally substituted  $-CH_2-C(=)-CH_2-$ ,  $-CH_2-CH(-)-CH_2-$ ,  $-CH=C(-)-$ ,  $-CH-C(=)-$  and  $-CH-CH(-)-$  groups.

According to a particularly preferred embodiment of the compound represented by formula (I), according to the present invention, the ring A is a six-membered ring to which X is covalently bonded by a single or a double bond to a carbon, a silicon, a phosphorus or a nitrogen atom represented by T, formula (I) of the light-stabilizer then becomes formula (II):



formula (II)

wherein,



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is exclusive of a nitrogen-nitrogen or nitrogen-oxygen bond, and wherein,

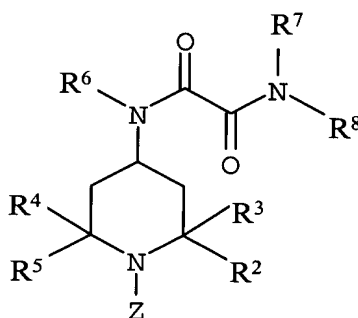
$R^2$  to  $R^5$  independently represent a substituted or unsubstituted C1 to C6 aliphatic group, most preferably  $R^2$  to  $R^5$  represent methyl or ethyl groups; Z is selected from the group consisting of

hydrogen, a substituted or unsubstituted aliphatic group, an acyl group, an oxy radical, a hydroxyl group, an alkoxy group and an acyloxy group; L is a divalent linking group linked to a carbonyl group; X and Y are independently selected from an oxygen and  $NR^6$ ,

wherein  $R^6$  is selected from the group consisting of hydrogen, a substituted or unsubstituted, saturated or unsaturated aliphatic group, a substituted or unsubstituted aromatic group, and a substituted or unsubstituted heteroaromatic group;  $R^1$  represents a non-aromatic moiety comprising at least two hydroxyl groups; and n and m independently represent 1 or 0.

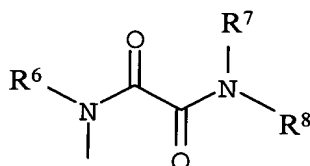
According to a particularly preferred embodiment of the compound represented by formula (II), according to the present invention, the atom T represents a carbon or a nitrogen atom.

In a particularly preferred embodiment the light-stabilizing compound is an oxalylamide derivative according to formula (III):



formula (III)

wherein,





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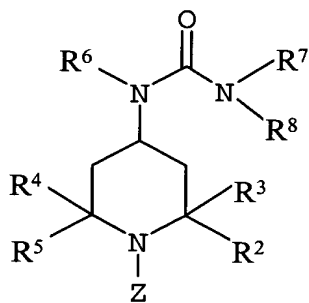
is exclusive of a nitrogen-nitrogen or nitrogen-oxygen bond, and wherein

$R^2$  to  $R^5$  independently represent a substituted or unsubstituted C1 to C6 aliphatic group, most preferably  $R^2$  to  $R^5$  represent methyl or ethyl groups;

Z is selected from the group consisting of hydrogen, a substituted or unsubstituted aliphatic group, an acyl group, an oxy radical, a hydroxyl group, an alkoxy group and an acyloxy group;  $R^6$  is

selected from the group consisting of hydrogen, a substituted or unsubstituted, saturated or unsaturated aliphatic group, a substituted or unsubstituted aromatic group, and a substituted or unsubstituted heteroaromatic group;  $R^7$  represents a non-aromatic moiety comprising at least two hydroxyl groups;  $R^8$  is selected from the group consisting of hydrogen, substituted or unsubstituted, saturated or unsaturated aliphatic group, a substituted or unsubstituted aromatic group.

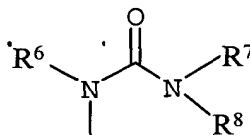
According to another particularly preferred embodiment the light-stabilizing compound is an urea derivative according to formula (IV):



formula (IV)

wherein,

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is exclusive of a nitrogen-nitrogen or nitrogen-oxygen bond, and wherein

$R^2$  to  $R^5$  independently represent a substituted or unsubstituted C1 to C6 aliphatic group, most preferably  $R^2$  to  $R^5$  represent methyl or ethyl groups;

Z is selected from the group consisting of hydrogen, a substituted or unsubstituted aliphatic group, an acyl group, an oxy radical, a hydroxyl group, an alkoxy group and an acyloxy group;  $R^6$  is selected from the group consisting of hydrogen, a substituted or unsubstituted, saturated or unsaturated aliphatic group, a substituted or unsubstituted aromatic group, and a substituted or unsubstituted heteroaromatic group;  $R^7$  represents a non-aromatic moiety comprising at least two hydroxyl groups;  $R^8$  is selected from the group consisting of hydrogen, substituted or unsubstituted, saturated or unsaturated aliphatic group, a substituted or unsubstituted aromatic group.

According to a preferred embodiment of the ink-recording material, according to the present invention,  $R^2$ ,  $R^3$ ,  $R^4$  and  $R^5$  are independently methyl or ethyl groups.

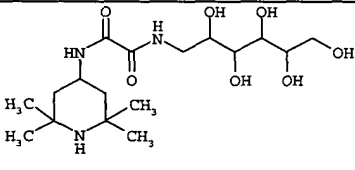
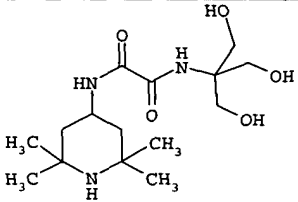
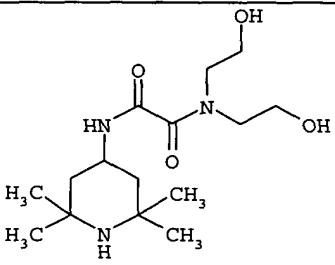
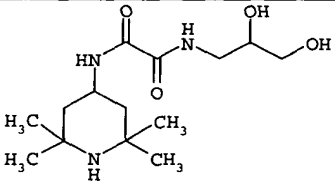
According to a preferred embodiment of the ink-recording material, according to the present invention, if substituents are present in  $R^1$ ,  $R^2$ ,  $R^3$ ,  $R^4$ ,  $R^5$ ,  $R^6$ ,  $R^7$ , and  $R^8$ , said substituents are selected from the group of hydroxyl groups, ether groups, carboxylic acid groups, ester groups, amide groups and amine groups.

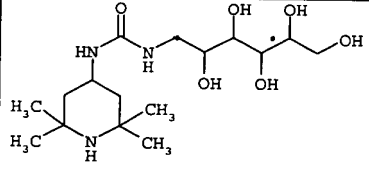
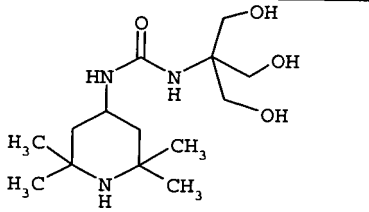
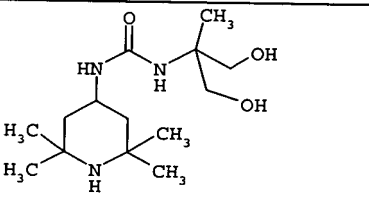
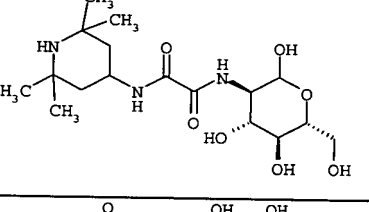
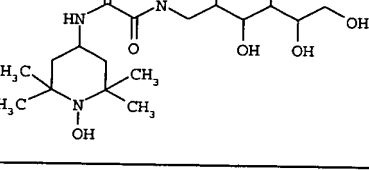
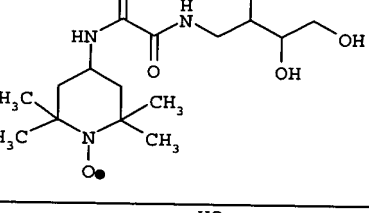
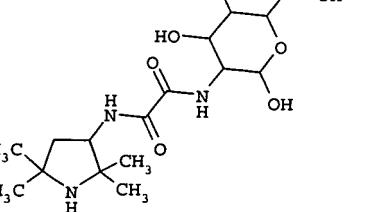
According to a preferred embodiment of the ink-recording material, according to the present invention,  $R^1$  is selected from the group consisting of optionally substituted polyhydroxy tetrahydro-pyrans, optionally substituted polyhydroxy tetrahydrofurans, polyhydroxy straight chain alkyl groups, polyhydroxy branched alkyl groups, polyhydroxy alkyl groups

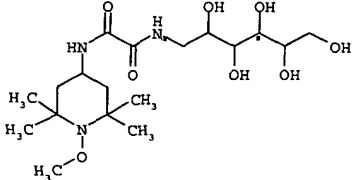
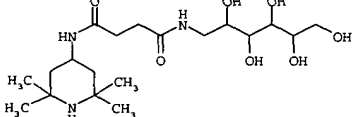
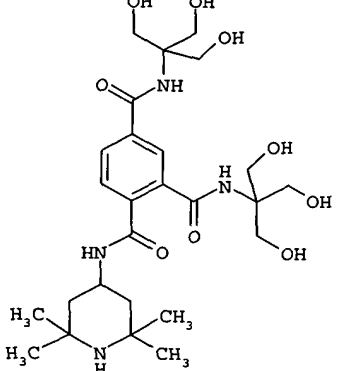
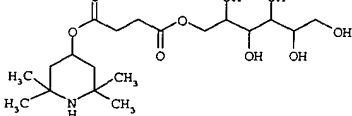
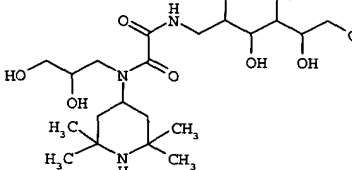
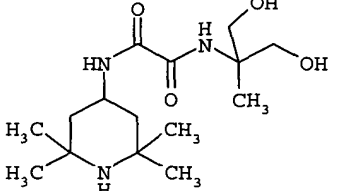
substituted with optionally substituted tetrahydropyran groups and polyhydroxy alkyl groups substituted with optionally substituted tetrahydrofuran groups.

According to a preferred embodiment of the ink-recording material, according to the present invention, the divalent linking group L is selected from the group consisting of  $-\text{CH}_2-\text{CH}_2-$ ,  $-\text{CH}=\text{CH}-$ ,  $-\text{CHR}^{10}-\text{CHR}^{11}-$  with  $\text{R}^{10}$  and  $\text{R}^{11}$  as part of a substituted or unsubstituted non-aromatic moiety,  $-\text{CR}^{10}=\text{CR}^{11}-$  with  $\text{R}^{10}$  and  $\text{R}^{11}$  as part of a substituted or unsubstituted aromatic group or a substituted or unsubstituted heteroaromatic group.

Useful light-stabilizing compounds used in accordance with the present invention are included in following list without being limited thereto.

	Stabilizer 1
	Stabilizer 2
	Stabilizer 3
	Stabilizer 4

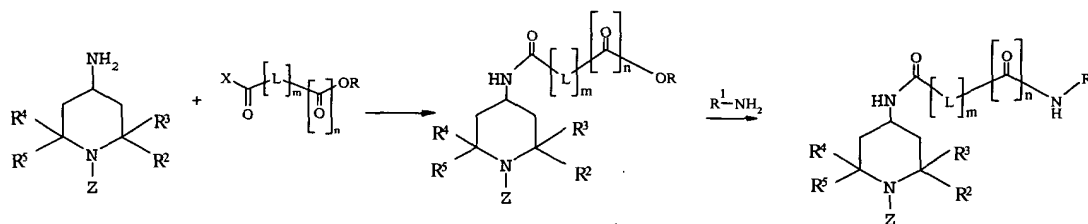
	Stabilizer 5
	Stabilizer 6
	Stabilizer 7
	Stabilizer 8
	Stabilizer 9
	Stabilizer 10
	Stabilizer 11

	Stabilizer 12
	Stabilizer 13
	Stabilizer 14
	Stabilizer 15
	Stabilizer 16
	Stabilizer 17

All compounds can be prepared according to well-known synthetic procedures. The synthesis of HALS-compounds used in accordance with the present invention is described with the generalized synthesis scheme below, wherein X is an electron

withdrawing group (for example a halogen atom or a methoxygroup), R represents an alkyl group and the other symbols are the same as defined for formula (I).

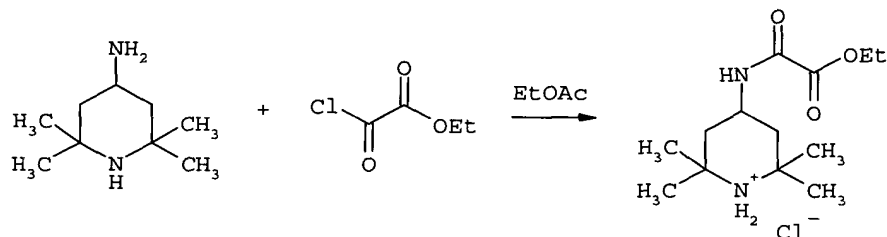
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The synthesis of HALS-compounds used in accordance with the present invention is described more in detail below, to illustrate the invention.

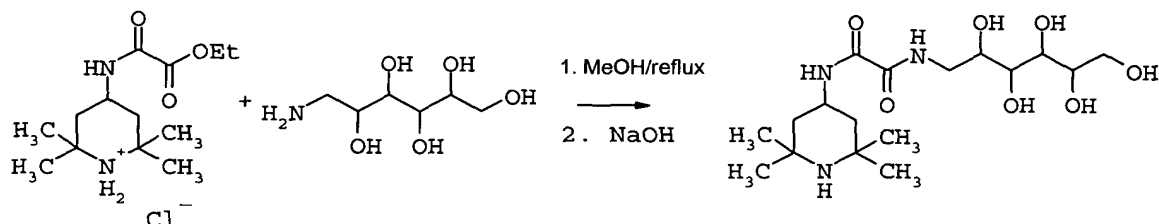
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#### The synthesis of stabilizer 1 :



87.5 ml (0.5 mol) of 4-amino-2,2,6,6-tetramethyl-piperidine was dissolved in 390 ml of ethyl acetate. 56 ml (0.5 mol) of ethyl oxalyl chloride was added dropwise, while keeping the temperature at 40°C. The reaction product precipitated from the medium. Upon complete addition, the reaction was allowed to continue for 2 hours at room temperature. The compound was isolated by filtration, washed twice with 50 ml of ethyl acetate and dried. 127 g (87%) of the oxalylamide chlorohydrate was isolated.

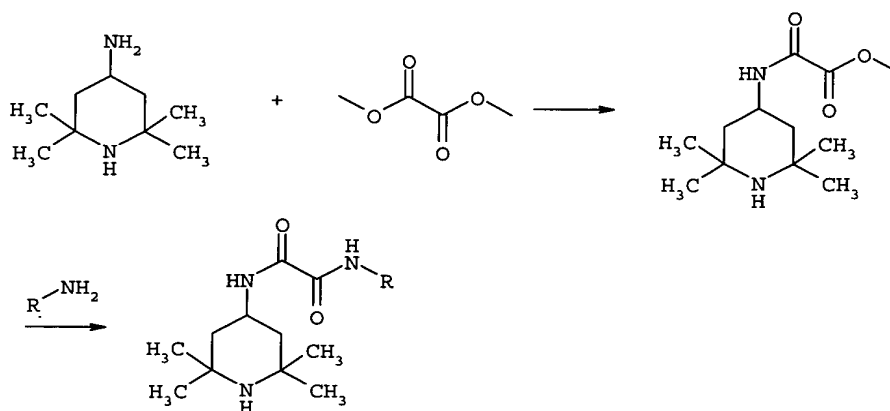
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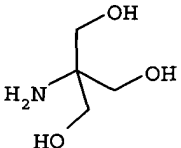
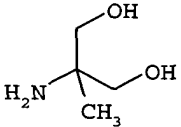
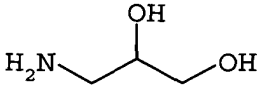
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32.2 g (0.11 mol) of the oxalyl amide chlorohydrate was dissolved in 275 ml of methanol. While stirring, 22 g (0.12 mol) of glucamine was added in 3 portions. The reaction mixture was refluxed for 7 hours. After cooling down to room temperature, the reaction mixture was concentrated to 150 ml. 300 ml of ethyl acetate was added and an oily product precipitated. The oily residue was treated several times with methanol/ethyl acetate 1/1 and finally with 300 ml of methyl-tert.butyl ether. On treating with methyl-tert. butyl ether, the product solidified and was isolated by filtration. 34.5 g of a crude chlorohydrate was isolated. Based on NP thin layer chromatography (eluent :  $\text{CH}_2\text{Cl}_2/\text{MeOH}/\text{NH}_3(\text{aq})$  : 70/24/6), it was judged that the HALS-oxalyl amide was insufficiently pure for evaluation. To remove the impurities, the chlorohydrate was converted into the free base with sodium hydroxide as follows. 34.5 g of the crude chlorohydrate was dissolved in 180 ml of water. 40 ml of 2N NaOH was added and the mixture was extracted 4 times with 180 ml of ethyl acetate. The aqueous fraction was isolated and water was removed under reduced pressure. The oily residue was treated with 180 ml of ethanol and the precipitated salts were removed by filtration. Ethanol was removed under reduced pressure and finally 17.7 g of compound 1 was isolated.

**The synthesis of stabilizers 2, 4 and 17:**

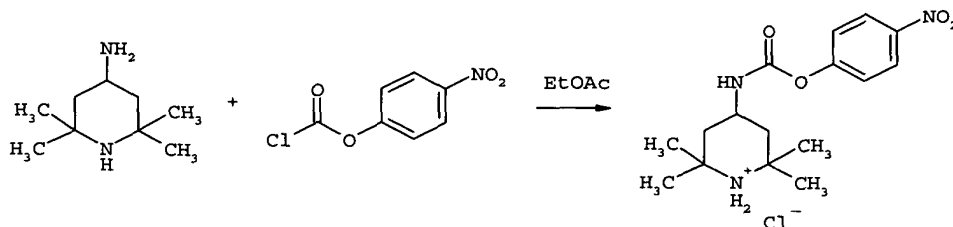


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R-NH <sub>2</sub>	Stabilizer
	Stabilizer 2
	Stabilizer 17
	Stabilizer 4

A solution of 15.6 g (0.1 mol) of 4-amino-2,2,6,6-tetramethylpiperidine in 30 ml of methanol was added dropwise to a solution of 14.26 g (0.12 mol) of dimethyl oxalate in 70 ml of methanol, while keeping the temperature at room temperature. The reaction was allowed to continue at room temperature for 2 hours. After two hours, 0.1 mol of the appropriate amine was added and the reaction was allowed to continue at room temperature for 16 hours. The precipitated side products were removed by filtration and methanol was removed under reduced pressure. The different stabilizers were purified by reversed phase chromatography on Kromasil C18 100 A 10 $\mu$ m, using a gradient elution, starting with MeOH/water, buffered with ammonium acetate at pH=7.6 (0.2M), 1/9 and ending with MeOH/water, buffered with ammonium acetate at pH=7.6 (0.2M), 38/62, using a flow rate of 150 ml/min. 11 g (33 %) of stabilizer 2, 9.4 g (30 %) of stabilizer 17 and 13.2 g (44%) of stabilizer 4 were isolated, partially as acetate salt.

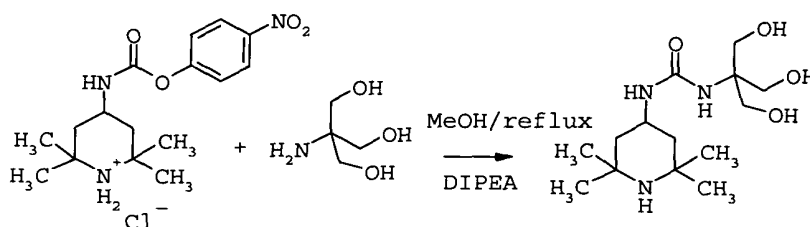
#### The synthesis of stabilizer 6 :





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3.4 g (16.8 mmol) of 4-nitrophenyl chloroformate was dissolved in 40 ml of ethyl acetate. The reaction mixture was cooled to 0°C and a solution of 2.39 g (15.3 mmol) of 4-amino-2,2,6,6-tetramethylpiperidine in 40 ml of ethyl acetate was added over 5 minutes. Immediately, the chlorohydrate of the 4-nitrophenyl-carbamate started to precipitate as a white solid. The reaction was allowed to continue for 1 hour at room temperature. The precipitated chlorohydrate was isolated by filtration, washed with ethyl acetate and dried. 5.0 g (91%) of the carbamate was isolated. The compound was sufficiently pure to be used without further purification.



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2 g (5.6 mmol) of the 4-nitrophenyl-carbamate was dissolved in 50 ml of methanol. 1.6 g (12.3 mmol) of DIPEA and 0.75 g (6.2 mmol) of tris(hydroxymethyl)aminomethane were added and the reaction mixture was refluxed for 3 hours. The solvent was removed under reduced pressure and stabilizer 6 was purified by reversed phase preparative column chromatography on Kromasil C18 100 A 10µm.

### Pigments

The pigment used is preferably an inorganic pigment, which can be chosen from neutral, anionic and cationic pigment types. Useful pigments include e.g. silica, talc, clay, hydrotalcite, kaolin, diatomaceous earth, calcium carbonate, magnesium carbonate, basic magnesium carbonate, aluminosilicate, aluminum trihydroxide, aluminum oxide (alumina), titanium oxide, zinc oxide, barium sulfate, calcium sulfate, zinc sulfide, satin white, alumina hydrate such as boehmite, zirconium oxide or mixed oxides.

Preferably, the pigment is a cationic type pigment selected from alumina hydrates, aluminum oxides, aluminum hydroxides, aluminum silicates, and cationically modified silicas.

A preferred type of alumina hydrate is crystalline boehmite, or  $\gamma$ -AlO(OH). Useful types of boehmite include, in powder form, DISPERAL, DISPERAL HP14 and DISPERAL 40 from Sasol, MARTOXIN VPP2000-2 and GL-3 from Martinswerk GmbH.; liquid boehmite alumina systems, e.g. DISPAL 23N4-20, DISPAL 14N-25, DISPERAL AL25 from Sasol. Patents on alumina hydrate include EP 500021, EP 634286, US 5,624,428, EP 742108, US 6,238,047, EP 622244, EP 810101, etc.. Useful cationic aluminum oxide (alumina) types include  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> types, such as NORTON E700, available from Saint-Gobain Ceramics & Plastics, Inc, and  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> types, such as ALUMINUM OXID C from Degussa; other aluminum oxide grades, such as BAIKALOX CR15 and CR30 from Baikowski Chemie; DURALOX grades and MEDIALOX grades from Baikowski Chemie, BAIKALOX CR80, CR140, CR125, B105CR from Baikowski Chemie; CAB-O-SPERSE PG003 trademark from Cabot, CATALOX GRADES and CATAPAL GRADES from from Sasol, such as PLURALOX HP14/150; colloidal Al<sub>2</sub>O<sub>3</sub> types, such as ALUMINASOL 100; ALUMINASOL 200, ALUMINASOL 220, ALUMINASOL 300, and ALUMINASOL 520 trademarks from Nissan Chemical Industries or NALCO 8676 trademark from ONDEO Nalco.

Other useful cationic inorganic pigments include aluminum trihydroxides such as Bayerite, or  $\alpha$ -Al(OH)<sub>3</sub>, such as PLURAL BT, available from Sasol, and Gibbsite, or  $\gamma$ -Al(OH)<sub>3</sub>, such as MARTINAL grades from Martinswerk GmbH, MARTIFIN grades, such as MARTIFIN OL104, MARTIFIN OL 107 and MARTIFIN OL111 from Martinswerk GmbH, MICRAL grades, such as MICRAL 1440, MICRAL 1500; MICRAL 632; MICRAL 855; MICRAL 916; MICRAL 932; MICRAL 932CM; MICRAL 9400 from JM Huber company; HIGILITE grades, e.g. HIGILITE H42 or HIGILITE H43M from Showa Denka K.K..

Another useful type of cationic pigment is zirconium oxide such as NALCO OOSS008 trademark of ONDEO Nalco, acetate stabilized ZrO<sub>2</sub>, ZR20/20, ZR50/20, ZR100/20 and ZRYS4 trademarks from Nyacol Nano Technologies.

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Useful mixed oxides are SIRAL grades from Sasol, colloidal metal oxides from Nalco such as Nalco 1056, Nalco TX10496, Nalco TX11678.

Another preferred type of inorganic pigment is silica which can be used as such in its anionic form or after cationic modification. Silica as pigment in ink receiving elements is disclosed in numerous old and recent patents, e.g. US 4,892,591, US 4,902,568, EP 373573, EP 423829, EP 487350, EP 493100, EP 514633, etc.. The silica can be chosen from different types, such as crystalline silica, amorphous silica, precipitated silica, fumed silica, silica gel, spherical and non-spherical silica. The silica may contain minor amounts of metal oxides from the group Al, Zr, Ti. Useful types include AEROSIL OX50 (BET surface area  $50 \pm 15 \text{ m}^2/\text{g}$ , average primary particle size 40 nm,  $\text{SiO}_2$  content > 99.8%,  $\text{Al}_2\text{O}_3$  content < 0.08%), AEROSIL MOX170 (BET surface area  $170 \text{ g/m}^2$ , average primary particle size 15 nm,  $\text{SiO}_2$  content > 98.3%,  $\text{Al}_2\text{O}_3$  content 0.3-1.3%), AEROSIL MOX80 (BET surface area  $80 \pm 20 \text{ g/m}^2$ , average primary particle size 30 nm,  $\text{SiO}_2$  content > 98.3%,  $\text{Al}_2\text{O}_3$  content 0.3-1.3%), or other hydrophilic AEROSIL grades available from Degussa-Hüls AG, which may give aqueous dispersions with a small average particle size (<500 nm).

Cationically modified silica can be prepared by following methods, without meaning to be limitative:

(1) subjecting silica to a surface treatment with an inorganic cationic compound such as particular metal oxides and oxyhydroxides, e.g. aluminum oxides, and alumina hydrates such as boehmite and pseudo-boehmite; a useful cationic inorganic compound to modify silica is pseudo-boehmite. Pseudo-boehmite is also called boehmite gel and is fine particulate alumina hydrate having a needle form. The composition thereof is generally represented by  $\text{Al}_2\text{O}_3 \cdot 1.5\text{-}2 \text{ H}_2\text{O}$  and differs from that of crystalline boehmite;

(2) by subjecting silica to a surface treatment with an organic compound having both an amino group or quaternary ammonium group thereof or a quaternary phosphonium group, and a functional group having reactivity to a silanol group on the surface of silica, such as aminoalkoxysilane or aminoalkyl glycidyl ether or isopropanol amine;

(3) by polymerisation of a cationic or amino functional monomer in the presence of a silica.

In an alternative embodiment the pigment may be chosen from organic particles such as polystyrene, polymethyl methacrylate, silicones, melamine-formaldehyde condensation polymers, urea-formaldehyde condensation polymers, polyesters and polyamides. Mixtures of inorganic and organic pigments can be used. However, most preferably the pigment is an inorganic pigment.

Mixtures of two or more pigments may be used. For obtaining glossy ink-receiving layers the particle size of the pigment should preferably be smaller than 500 nm. In order to obtain a porous glossy layer which can serve as an ink-receiving layer for fast ink uptake the pigment/binder ratio should be at least 4. Only at these high ratios of the pigments longer able to fill up all pores and voids created by the pigments in the coating. To achieve a sufficient porosity of the coating for fast ink uptake the pore volume of these highly pigmented coatings should be higher than 0.1 ml/g of coated solids. This pore volume can be measured by gas adsorption (nitrogen) or by mercury diffusion.

Cationic substance acting as a mordant

Cationic substances acting as a mordant increase the capacity of the layer for fixing and holding the dye of the ink droplets. A particularly suited compound is a poly(diallyldimethylammonium chloride) or, in short, a poly(DADMAC). These compounds are commercially available from several companies, e.g. Aldrich, Nitto Boseki Co., Clariant, BASF and EKA Chemicals.

Other useful cationic compounds include DADMAC copolymers such as Nalco, CIBA, trademark of Nitto Boseki Co., such as copolymers with acrylamide, e.g. NALCO 1470 trade mark of ONDEO Nalco or PAS-J-81, trademark of Nitto Boseki Co., such as copolymers of DADMAC with acrylates, such as Nalco 8190, trademark of ONDEO Nalco; copolymers of DADMAC with SO<sub>2</sub>, such as PAS-A-1 or PAS-92, trademarks of Nitto Boseki Co., copolymer of DADMAC with maleic acid, e.g. PAS-410, trademark of Nitto Boseki Co., copolymer of DADMAC with diallyl(3-chloro-2-hydroxypropyl)amine

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hydrochloride, eg. PAS-880, trademark of Nitto Boseki Co., dimethylamine-epichlorohydrine copolymers, e.g. Nalco 7135, trademark of ONDEO Nalco or POLYFIX 700, trade name of Showa High Polymer Co.; other POLYFIX grades which could be used are POLYFIX 5 601, POLYFIX 301, POLYFIX 301A, POLYFIX 250WS, and POLYFIX 3000 ; NEOFIX E-117, trade name of Nicca Chemical Co., a polyoxyalkylene polyamine dicyanodiamine, and REDIFLOC 4150, trade name of EKA Chemicals, a polyamine; MADAME (methacrylatedimethylaminoethyl = dimethylaminoethyl methacrylate) or MADQUAT 10 (methacryloxyethyltrimethylammonium chloride) modified polymers, e.g. ROHAGIT KL280, ROHAGIT 210, ROHAGIT SL144, PLEX 4739L, PLEX 3073 from Röhm, DIAFLOC KP155 and other DIAFLOC products from Diafloc Co., and BMB 1305 and other BMB products from EKA chemicals; cationic epichlorohydrin adducts such as POLYCUP 171 15 and POLYCUP 172, trade names from Hercules Co.; from Cytec industries : CYPRO products, e.g. CYPRO 514/515/516, SUPERFLOC 507/521/567; cationic acrylic polymers, such as ALCOSTAT 567, trademark of CIBA, cationic cellulose derivatives such as CELQUAT L-200, H-100, SC-240C, SC-230M, trade names of Starch & Chemical 20 Co., and QUATRISOFT LM200, UCARE polymers JR125, JR400, LR400, JR30M, LR30M and UCARE polymer LK; fixing agents from Chukyo Europe: PALSET JK-512, PALSET JK512L, PALSET JK-182, PALSET JK-220, WSC-173, WSC-173L, PALSET JK-320, PALSET JK-320L and PALSET JK-350; polyethyleneimine and copolymers, e.g. LUPASOL, trade name 25 of BASF AG; triethanolamine-titanium-chelate, e.g. TYZOR, trade name of Du Pont Co.; copolymers of vinylpyrrolidone such as VIVIPRINT 111, trade name of ISP, a methacrylamido propyl dimethylamine copolymer; with dimethylaminoethylmethacrylate such as COPOLYMER 845 and COPOLYMER 937, trade names of ISP; with 30 vinylimidazole, e.g. LUVIQUAT CARE, LUVITEC 73W, LUVITEC VPI55 K18P, LUVITEC VP155 K72W, LUVIQUAT FC905, LUVIQUAT FC550, LUVIQUAT HM522, and SOKALAN HP56, all trade names of BASF AG; polyamidoamines, e.g. RETAMINOL and NADAVIN, trade marks of Bayer AG; phosphonium compounds such as disclosed in EP 609930 and other 35 cationic polymers such as NEOFIX RD-5, trademark of Nicca Chemical Co.

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The ink receiving layer, and an optional auxiliary layer, such as a backing layer for anti-curl purposes, may further contain well-known conventional ingredients, such as surfactants serving as coating aids, hardening agents, plasticizers, whitening agents and matting agents.

### Surfactants

Surfactants may be incorporated in the layers of the recording element of the present invention. They can be any of the cationic, anionic, amphoteric, and non-ionic ones as described in JP-A 62-280068 (1987). Examples of the surfactants are N-alkylamino acid salts, alkylether carboxylic acid salts, acylated peptides, alkylsulfonic acid salts, alkylbenzene and alkyl-naphthalene sulfonic acid salts, sulfosuccinic acid salts,  $\alpha$ -olefin sulfonic acid salts, N-acylsulfonic acid salts, sulfonated oils, alkylsulfonic acid salts, alkylether sulfonic acid salts, alkylallylethersulfonic acid salts, alkylamidesulfonic acid salts, alkylphosphoric acid salts, alkylether-phosphoric acid salts, alkylallyletherphosphoric acid salts, alkyl and alkylallylpolyoxyethylene ethers, alkylallylformaldehyde condensed acid salts, alkylallylethersulfonic acid salts, alkylamide-sulfonic acid salts, alkylphosphoric acid salts, alkyletherphosphoric acid salts, alkylallyletherphosphoric acid salts, alkyl and alkylallylpolyoxyethylene ethers, alkylallylformaldehyde condensed polyoxyethylene ethers, blocked polymers having polyoxypropylene, polyoxyethylene polyoxypropyl-alkylethers, polyoxyethyleneether of glycolesters, polyoxyethylene-ether of sorbitanesters, polyoxyethyleneether of sorbitolesters, poly-ethyleneglycol aliphatic acid esters, glycerol esters, sorbitane esters, propyleneglycol esters, sugaresters, fluoro C2-C10 alkyl-carboxylic acids, disodium N-perfluorooctanesulfonate, sodium 3-(fluoro-C6-C11-alkyloxy)-1-C3-C4 alkyl sulfonates, sodium 3-( $\omega$ -fluoro-C6-C8alkanoyl-N-ethylamino)-1-propane sulfonates, N-[3-(perfluorooctanesulfonamide)-propyl]-N,N-dimethyl-N-carboxymethylene ammonium betaine, fluoro-C11-C20 alkylcarboxylic acids,

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perfluoro-C7-C13-alkyl-carboxylic acids, perfluorooctane sulfonic acid diethanolamide, Li, K and Na perfluoro-C4-C12-alkyl sulfonates, N-propyl-N-(2-hydroxyethyl)per-fluorooctane sulfonamide, perfluoro-C6-C10-alkylsulfonamide-propyl-sulfonyl-glycinates, bis-(N-perfluorooctylsulfonyl-N-ethanolamino-ethyl)phosphonate, mono-perfluoro C6-C16 alkyl-ethyl phosphonates and perfluoroalkyl-betaine.

Useful cationic surfactants include N-alkyl dimethyl ammonium chloride, palmityl trimethyl ammonium chloride, dodecyldimethylamine, tetradecyldimethylamine, ethoxylated alkyl guanidine-amine complex, oleamine hydroxypropyl bistrimonium chloride, oleyl imidazoline, stearyl imidazoline, cocamine acetate, palmitamine, dihydroxyethylcocamine, cocotrimonium chloride, alkyl polyglycolether ammonium sulphate, ethoxylated oleamine, lauryl pyridinium chloride, N-oleyl-1,3-diaminopropane, stearamidopropyl dimethylamine lactate, coconut fatty amide, oleyl hydroxyethyl imidazoline, isostearyl ethylimidonium ethosulphate, lauramidopropyl PEG-dimoniumchloride phosphate, palmityl trimethylammonium chloride, and cetyltrimethylammonium bromide.

Especially useful are the fluorocarbon surfactants as described in e.g. US 4,781,985, having a structure of:  $F(CF_2)_4-9CH_2CH_2SCH_2CH_2N+R_3X^-$  wherein R is a hydrogen or an alkyl group; and in US 5,084,340, having a structure of:  $CF_3(CF_2)_mCH_2CH_2O(CH_2CH_2O)_nR$  wherein  $m = 2$  to 10;  $n = 1$  to 18; R is hydrogen or an alkyl group of 1 to 10 carbon atoms. These surfactants are commercially available from DuPont and 3M. The concentration of the surfactant component in the ink-receiving layer is typically in the range of 0.1 to 2 %, preferably in the range of 0.4 to 1.5 % and is most preferably 0.75 % by weight based on the total dry weight of the layer.

#### Plasticizers

The ink-receiving layer and the optional auxiliary layer(s) may also comprise a plasticizer such as ethylene glycol, diethylene glycol, propylene glycol, polyethylene glycol, glycerol monomethylether, glycerol monochlorohydrin, ethylene carbonate,

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propylene carbonate, urea phosphate, triphenylphosphate, glycerol-monostearate, propylene glycol monostearate, tetramethylene sulfone, n-methyl-2-pyrrolidone, n-vinyl-2-pyrrolidone.

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### Crosslinking agents

There are a vast number of known crosslinking agents - also known as hardening agents - that will function to crosslink film forming binders. Hardening agents can be used individually or in  
10 combination and in free or in blocked form. A great many hardeners, useful for the present invention, are known, including formaldehyde and free dialdehydes, such as succinaldehyde and glutaraldehyde, blocked dialdehydes, active esters, sulfonate esters, active halogen compounds, isocyanate or blocked  
15 isocyanates, polyfunctional isocyanates, melamine derivatives, s-triazines and diazines, epoxides, active olefins having two or more active bonds, carbodiimides, zirconium complexes, e.g. BACOTE 20, ZIRMEL 1000 or zirconium acetate, trademarks of MEL Chemicals, titanium complexes, such as TYZOR grades from DuPont, isoxazolium  
20 salts substituted in the 3-position, esters of 2-alkoxy-N-carboxy-dihydroquinoline, N-carbamoylpyridinium salts, hardeners of mixed function, such as halogen-substituted aldehyde acids (e.g. mucochloric and mucobromic acids), onium substituted acroleins and vinyl sulfones and polymeric hardeners, such as dialdehyde  
25 starches and copoly(acroleinmeth-acrylic acid), and oxazoline functional polymers, e.g. EPOCROS WS-500, and EPOCROS K-1000 series, and maleic anhydride copolymers, e.g. GANTREZ AN119.

In the practice of this invention boric acid is a preferred crosslinker.

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The present invention will now be illustrated by the following examples without however being limited thereto.

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### EXAMPLES

#### Exempl 1



### Preparation of the coating solution

To apply the light-stabilizing agent to ink jet print media, a coating liquid was prepared by adding 25 parts by solid weight of a 10% aqueous solution of the light-stabilizing compounds ST-1, ST-17, and ST-2 respectively (corresponding to stabilizers 1, 17 and 2 from the compounds list in the Detailed Description section above) to 170 parts of water.

### Coating and evaluation of the coated samples

The coating solution was applied to a glossy porous medium (Agfajet Universal Instant Dry Photograde Paper Glossy) as the basecoat. In order to change the concentration of the light-stabilizing additive, the thickness of the applied coating solution was varied. The application was performed by means of a doctor blade coater. The comparative samples were obtained by simply applying the aqueous solution to the recording media without the stabilizing agent.

After the media were allowed to dry at room temperature for 24 hours, color patches with 50% and 100% ink of cyan, magenta, yellow and black were printed by means of a printer HP970Cxi (trademark of Hewlett-Packard). The black patches were obtained by printing cyan, magenta and yellow. The light-fastness was evaluated by measuring the relative optical density loss of the printed samples after being exposed to light in a fade-o-meter, XENOTEST 150 (trademark: Original Hanau) with 180 kLux during 16 hours. The results are summarised in table 1.

TABLE 1 : comparison of optical density loss of printed media (50% patches) treated with and without light-stabilizing additive:

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Color	Relative optical density loss (%)			
	Without ST	With 1.0 g/m <sup>2</sup> of ST-1	With 1.0 g/m <sup>2</sup> of ST-17	With 0.5 g/m <sup>2</sup> of ST-2
Yellow	30.8	23.6	13.5	21.8
Magenta	41.9	19.4	13.6	20.9
Cyan	41.2	32.1	28.9	33.3
Black	36.2	18.6	15.5	20.0

As can be seen from the table, the porous materials  
5 impregnated with the stabilizers ST-1, ST-17, and ST-2 show an  
significant improvement in light-fastness.

#### Example 2

10 Ink jet recording media (invention and comparison) were  
prepared by coating on a resin-coated paper an ink receiving layer  
which composition is shown in table 2. The coating weight of the  
inorganic pigment was 30.0 g/m<sup>2</sup>. On top of this layer, a gloss  
improving layer is coated containing 100 parts of the commercial  
15 boehmite DISPERAL HP 14/2, from Sasol Co., a 25 % dispersion in  
water, 2 parts of a polyvinyl alcohol GOHSEFIMER K210 from Nippon  
Goshei Co., 0.2 parts of boric acid, and 0.8 parts of  
cetyltrimethylammonium bromide. The coating thickness of the top  
layer was chosen to achieve a pigment coating weight of 5 g/m<sup>2</sup>.  
20 The invention samples contained 20 parts of the stabilizing agents  
ST-1 and ST-2 respectively in the top layer, the comparative  
sample did not.

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TABLE 2

Compositions	
Alumina (Cab-o-Sperse PG003: 40%, supplied by Cabot Corp.)	704.3 g
Polyvinyl alcohol (Gohsefimer K210, supplied by Nippon Gohsei)	18.3 g
Pseudo-boehmite DISPERAL P3 (trade name of Sasol Co.)	28.2 g
Boric acid	1.8 g
Deionised water	477.5 g

Color patches with 50% and 100% ink of cyan, magenta, yellow  
5 and black were printed on both samples by means of a printer  
HP970Cxi (trademark of Hewlett-Packard). The light-fastness was  
evaluated by measuring the relative optical density loss of the  
printed samples after being exposed to light in a fade-o-meter,  
XENOTEST 150 (trademark of Original Hanau) with 180 kLux during 16  
10 hours.

The relative loss (%) in density of the four 50% color  
patches together, due to the light fading are shown in table 3.

TABLE 3

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Color	Comp. Sample without ST	Inv. Sample with ST-1	Inv. sample with ST-2
Yellow	42.1	39.6	37.5
Magenta	80.3	59.7	56.3
Cyan	49.0	44.2	43.4
Black	54.4	41.4	45.0

As can be seen, the color stability, especially for the  
magenta ink and the black color (obtained from cyan, magenta and  
yellow ink), was much improved when compounds ST-1 and ST-2 were  
20 introduced in the top layer.

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Having described in detail preferred embodiments of the current invention, it will now be apparent to those skilled in the art that numerous modifications can be made therein without departing from the scope of the invention as defined in the appending claims.